

# Structural, Optical and Mechanical Properties of Pure and Pb<sup>2+</sup>Doped L-Threonine Manganese Acetate Single Crystals

N. Imthiyas Ahamed, D. Benny Anburaj, G. Nedunchezhian

Abstract: Single crystals of Pure L-Threonine manganese acetate (LTMA) and Lead (Pb2+) doped LTMA have been grown by the slow evaporation method grown successfully by slow evaporation method at 300C Temperature. Single crystal XRD study determines the crystal system and unit cell parameters, and it carried out volume variations. Powder XRD pattern confirm that there is change in the lattice points of pure and doped material. The prominent peaks confirm the positions of lattice points and crystal planes. EDAX analysis qualitatively ignored that the presence of Chemical composition in both crystals. The active functional groups of the crystal points are analyzed by FTIR spectrum. Optical property of the crystal studied by UV-Vis-NIR, shows that transmittance in the range of 267-1100 nm for pure LTMA and 200 -1100 nm for Pb2+ doped LTMA and energy band gaps also determined. The Hardness of the both crystals is studied by Vicker's Micro hardness analysis.

Keywords: Single crystal XRD study, Optical property, EDAX analysis, Mechanical property

#### I. INTRODUCTION

At present lot of the exploration work done on the synthesizing and portrayal of semi natural crystal. Because of their wide optical applications, contrasted with different materials, amino corrosive blended natural crystal was intrigued to orchestrated [1]. The search materials has qualifies for the discovery of numerous natural NLO materials with high nonlinear natures and photonic applications [2–4] and [5-11]. Authentically, it can utilize in uses of predominant quality precious stones [12, 13]. Like the semi natural precious stones, amino corrosive based semi natural gems were likewise has acceptable optical and non-direct optical natures. This single crystal can likewise be

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developed from watery answer for the improved hardness and the warm steadiness. In this reference, unadulterated L-Threonine single gem and L-Threonine based Lithium Chloride (LTLC), Calcium Chloride (LTCC), Cadmium chloride (LTCC), Manganese chloride (LTMC) single precious crystal are developed and its characters likewise examined [14-18]. Afterward, L-Threonine Sulfate crystal like Lithium Sulfate (LTLS), Potassium Sulfate (LTKS), Zinc Sulfate (LTZS) and Copper Sulfate (LTCS) single precious crystal are developed and considered [19-22]. At that point, L-Threonine Cadmium Acetate (LTCA) and L-Threonine Zinc Acetate (LTZA) single precious crystal are developed [23-24]. In proceeds, this work, finished with development on unadulterated L-Threonine Manganese Acetate single crystal (LTMA) and the Lead doped L-Threonine Manganese Acetate (Pb2+ doped LTMA) novel single crystals were effectively developed by moderate vanishing technique. The different portrayals have been done and those properties of both crystals were detailed.

#### II. EXPERIMENTAL SECTION

#### A. Synthesis & Crystal growth

AR grade chemicals of L-Threonine and Manganese acetate tetra hydrate in the ratio of 1:1M taken in 10ml distilled water and 0.2M of (AR grade) lead acetate (PbA) added in 1:0.8 M of L-Threonine and Manganese Acetate solution taken in 10ml distilled water separately. Both prepared saturated solutions was filtered and then housed in a dust free atmosphere. The pure and Pb<sup>2+</sup> doped LTMA single crystals were successfully grown at room temperature with repeated recrystallization by slow evaporation method. After a period of 20-25 days, harvested single crystals of  $10\times3\times2$  mm<sup>3</sup> size pure LTMA single crystal and  $14\times4\times3$  mm<sup>3</sup> size of Pb<sup>2+</sup> doped LTMA were grown as shown in Fig.1.

Chemical formula:

i) Pure LTMA single crystal C<sub>4</sub>H<sub>9</sub>NO<sub>3</sub>+(CH<sub>3</sub>COO)<sub>2</sub>Mn.4H<sub>2</sub>O

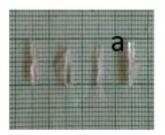
 $\rightarrow$  Mn[C<sub>4</sub>H<sub>9</sub>NO<sub>3</sub>]<sup>+</sup> . 2(CH<sub>3</sub>COO)<sup>-</sup>+4H<sub>2</sub>O  $\uparrow$  (1)

ii) 0.2M Pb<sup>2+</sup> doped LTMA single crystal  $C_4H_9NO_3+0.8M(CH_3COO)_2Mn.4H_2O$ 

+0.2M(CH<sub>3</sub>COO)<sub>2</sub> Pb

 $\rightarrow$  Mn<sub>0.8</sub>[C<sub>4</sub>H<sub>9</sub>NO<sub>3</sub>]Pb<sub>0.2</sub>)+.2(CH<sub>3</sub>COO)<sup>-</sup>+4H<sub>2</sub>O  $\uparrow$  (2)





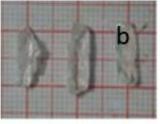


Fig.1. As grown (a) pure LTMA and (b) Pb<sup>2+</sup> doped LTMA single crystals

#### III. RESULT AND DISCUSSION

#### A. Single Crystal X – Ray Diffraction Analysis

In order to determine the Crystal system, the grown single crystals of pure LTMA and Pb $^{2+}$  doped LTMA subjected into Single Crystal XRD analysis. SCXRD analysis carried out using Nonius CAD4/MACH3 Single Crystal X-ray diffractometer with MoK $\alpha$  ( $\alpha = 0.71073$  Å) source lies that pure L-Threonine Manganese Acetate (LTMA) and Pb $^{2+}$  doped LTMA become Orthorhombic system with different unit cell parameters as shown in Table 1. Previous reports of similar base material L-Threonine (L-T)[14] and Cadmium acetate doped L-Threonine single crystal (LTCA)[24] are compared in this Table I. This result proved that pure LTMA and Pb $^{2+}$  doped LTMA also satisfy the previous same reports in this regard.

Table- I: Lattice Parameters of Pure and  $Pb^{2+}$  Doped LTMA

211.21			
Parameter	Pure LTMA	Pb <sup>2+</sup> doped LTMA	
a	5.106(19) Å	5.147(3) Å	
b	7.721(19) Å	7.731(2) Å	
b	13.45(4) Å	13.590(10) Å	
α=β=γ	90°	90°	
Volume	530(4) Å <sup>3</sup>	540.8(4) Å <sup>3</sup>	
Space group	P <sub>212121</sub>	P <sub>212121</sub>	
Crystal			
System	Orthorhombic	Orthorhombic	

#### B. Powder X – ray diffraction analysis

To study the Powder X-ray diffraction analysis recorded is to determine the lattice point and repeated regular atoms illustrated. This pattern results that different peaks reports the strength of atoms of compositions regarding the reaching the height of peaks are shown in Fig.2. Miller indices estimated by powder V1.0 software along with 2 Theta values of LTMA crystal is given in Table II. The different peaks confirm the Powder XRD pattern L-Threonine Manganese Acetate single crystal. All the peaks differ from the peaks of pure L-Threonine single crystal [14]. The first peak start at 3.93 degree and its plane calculated [111] and second maximum reaches as 345 a.u intensity in angle 22.2 degree, its plane calculated as [411]. The continual peaks reveal at the angles 25.6, 26.6, 36, 38.4 and 44.9 in the respective peaks of plane 422, 431, 444, 552 and 661. Fig.2 proved that the pure

LTMA consist, number of different planes with different lattice points. It concluded that no one single element result as different peaks, therefore this material contain greater single element. In this Pb<sup>2+</sup> doped LTMA single crystal, the difference in peak powder XRD Pattern are determined and its all the different hkl plane values are calculated and tabulated in the Table III.

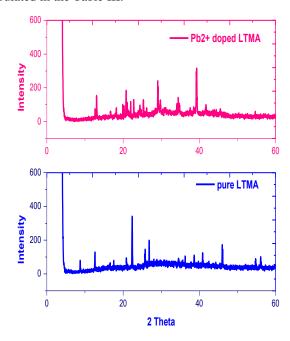


Fig. 2: Powder XRD patterns of pure and Pb<sup>2+</sup> doped LTMA

Table-II: 20 Vs Intensity Values of LTMA Crystal

2θ	Intensity	hkl
3.93	1178	111
12.5	134	211
17.6	100	222
22.2	345	411
25.6	154	422
26.6	203	431
36.0	108	444
38.4	113	552
44.9	176	661

Table-III: 20 Vs Intensity Values of Pb2+ Doped LTMA

2θ	Intensity	hkl
3.93	1089	111
13.7	155	216
16.6	62.8	(-3)66
18.2	85	555
20	103	581
20.8	189	6(-1)8
22.6	129	749

#### C. Energy Dispersive analysis (EDAX)

Energy dispersive X-ray analysis (EDAX) illustrates the EDAX spectrum of LTMA crystal using JEOL company (JSM-6701 F, SEM). The presence of Carbon, Nitrogen, Oxygen and Manganese in LTMA single crystal was obtained as shown in Fig. 3.





Due to the inclusion of acetic acid, Carbon & Oxygen has the maximum peaks. Manganese places are clearly displayed in EDAX Spectrum. The percentage of elements recorded and showed that the compounds contained the elements as: Carbon (C), Nitrogen (N), Oxygen (O) and Manganese (Mn). In Table IV, displays the weight, percentage of compound placed in crystal as experimental report. It proves the purity and exacts of the crystal.

In this Pb<sup>2+</sup> doped LTMA single crystal, doped composition are clearly indicated in the EDAX Spectrum as shown in Fig.4. In which the peaks of Carbon (C), Nitrogen (N), Oxygen (O), Manganese (Mn) and dopant lead (Pb) are well defined. In this determination the weight of the elements are given as experimental report Table V.

The atomic mass percentages of C, N, O, Mn in Pure material and inclusion of Pb also experimentally prove in the composition and grown success of raw material.

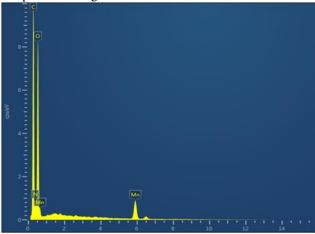


Fig. 3 EDAX Spectrum of pure LTMA

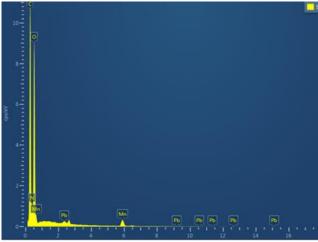


Fig. 4 EDAX Spectrum of Pb<sup>2+</sup> doped LTMA crystals

#### Table- IV: EDAX values of pure LTMA

Element	Line Type	Wt%	Atomic %
С	K series	41.51	49.7
N	K series	6.76	6.94
0	K series	46.79	42.06
Mn	K series	4.94	1.29
Total:		100	100

Table-V: EDAX values of Pb2+ doped LTMA crystal

Element	Line Type	Wt%	Atomic %
С	K series	40.33	47.47
N	K series	9.14	9.22
0	K series	48.55	42.9
Mn	K series	1.45	0.37
Pb	M series	0.54	0.04
Total:		100	100

## C. Fourier Transform Infra-Red a nalysis

The FTIR spectrum of Pure and Pb2+ doped LTMA crystals was recorded as shown in Fig. 5. The presence of NH<sub>3</sub><sup>+</sup> is easily identified in the FT-IR spectrum by the broad intense band with the absorption at 3169.25cm<sup>-1</sup> corresponding to asymmetric stretching mode of NH<sub>3</sub><sup>+</sup>. The NH<sub>3</sub><sup>+</sup> symmetric stretching band appears at 3029.80 cm<sup>-1</sup>. The NH<sub>3</sub><sup>+</sup> symmetric and asymmetric bending mode appears at 1417.94 cm<sup>-1</sup> and 1629.28 cm<sup>-1</sup>. The CH symmetric deformation appears in 1346.04 cm<sup>-1</sup> respectively. NH<sub>3</sub><sup>+</sup> Rocking, C-N Rocking, C-C Rocking and C-C-N Rocking appear in 1113.14 cm<sup>-1</sup>.1040.72 cm<sup>-1</sup>, 932.53 cm<sup>-1</sup> and 871.20 cm<sup>-1</sup>. 769.77 cm<sup>-1</sup> indicates COO Bending, 701.35 cm<sup>-1</sup> and 560.11 cm<sup>-1</sup> indicates the COO wagging vibration and COO<sup>-</sup> Rocking Deformation respectively. NH<sub>3</sub><sup>+</sup> Bending appears at 489.82 cm<sup>-1</sup>.The OCO rocking frequency found out at 445.18 cm<sup>-1</sup>.

In the Pb<sup>2+</sup> doped LTMA single crystal FT-IR spectrum shows small changes in all the peaks with examples of NH<sub>3</sub><sup>+</sup> symmetric stretching and asymmetric bands, CH symmetric deformation and CH<sub>3</sub> group, NH<sub>3</sub><sup>+</sup> and CH<sub>3</sub><sup>+</sup> rocking frequencies absorptions and COO- Bending groups also. Especially at stretched symmetric and asymmetric several peaks are large variations in structure. The OCO rocking frequency found out at 445.89 cm<sup>-1</sup>. These are confirming incorporation of the new element such as Pb<sup>2+</sup> ion in pure LTMA. The compared readings of the Pb<sup>2+</sup> doped LTMA single crystal tabulated in Table VI.



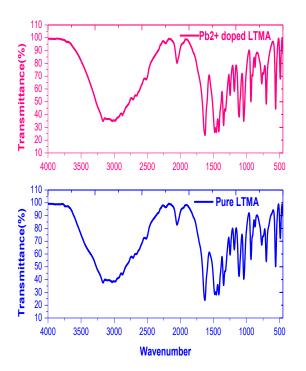


Fig. 5. FTIR graph of the pure LTMA and Pb<sup>2+</sup> doped LTMA crystals

Table-VI: FTIR Assessment of Pure LTMA and Pb<sup>2+</sup> doped LTMA

uopeu LIVIA			1
	Wavenumber		
Pure			
L-Threonine		Pb²+ doped	
(14)	Pure LTMA	LTMA	Assignment
			$NH_3^+$
			asymmetric
-	3169.25	3168.64	stretching
			$NH_3^+$
			symmetric
-	3029.8	3028.16	stretching
			$\mathrm{NH_{3}^{+}}$
			asymmetric
1625.99	1629.28	1627.41	deformation
			$NH_3^+$
			symmetric
1417	1417.94	1417.77	deformation
			CH
			symmetric
1346.48	1346.04	1346.45	deformation
	1113.14	1111.96	NH <sub>3</sub> <sup>+</sup> Rocking
1040.52	1040.72	1041.66	C-N Rocking
931.5	932.53	932.77	C-C Rocking
			C-C-N
871.09	871.2	871.29	Rocking
767.9	769.77	769.05	COO Bending
			COO-wagging
701.52	701.35	701.62	vibration
			COO-Rocking
560.19 C	560.11	560.23	Deformation
489.64	489.82	489.3	NH <sub>3</sub> <sup>+</sup> Bending
			In plan OCO
-	44518	445.89	Rocking

#### E. Optical Studies

#### a. Transmittance Studies

The transmittance spectra of pure LTMA recorded in the range 190-1200 nm using Lambda 35 spectrometer. The Optical transmittance spectra of LTMA are shown in Fig.6. It reveals that there is no absorption peak in the range of 267 nm

to 1100 nm. It can be seen from the transmission curve that below 300nm the transmittance of the grown crystal LTMA slightly decreases [25]. Variation in the transmittance may be due to the presence of manganese acetate. Very low absorbance in the entire visible region would be attributed to the delocalization of electronic cloud through charge transfer. For Pb<sup>2+</sup> doped LTMA transmittance start to transmit from 200nm to 1100nm. This change in transmittance of pure LTMA and the Pb<sup>2+</sup> doped LTMA clearly separated reason of lead (Pb).

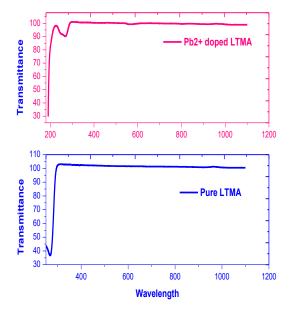


Fig. 6. The UV-Visible Transmittance Spectra of pure and  $Pb^{2+}$  doped LTMA crystals

#### **b.** Absorption Studies

The absorption spectrum of grown crystal analyzed in the range 190-1200 nm. Fig.7 shown, there is no change from the transmittance spectra. The absence of absorption in the visible region clearly indicates that the grown crystal can be used for photonic applications. The absence of absorption spectrum concluded the good energy band gap value. By knowing optical constants of a material, examine the potential of the material for photonics applications. The optical absorption coefficient ( $\alpha$ ) calculated by the relation

$$\alpha = (1/t) \log(1/T) \tag{3}$$

Where, t is Thickness of the material and T is Transmittance.

The band gap of the crystal was estimated by Tauc's relation:

$$\alpha h v = A(h v - E_g)^n \tag{4}$$

Where,  $E_g$  is the optical band gap of the crystal and A is a constant. So, the energy band gap of grown crystal was determined from the Fig. 7, as 4.3eV and the energy band gap  $Pb^{2+}$  doped LTMA is 6.3eV.





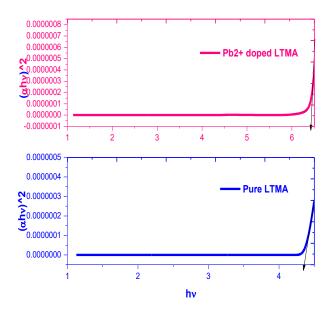


Fig. 7. Energy band gap of pure LTMA and Pb<sup>2+</sup> doped LTMA crystals

#### F. Vicker's Micro Hardness Study (MHD)

The micro hardness study was carried out to determine the mechanical strength of the grown crystals using HMT 2T, SHIMADZU Vickers micro hardness tester. The indentation marks were made on the surface of the both crystals at room temperature by applying load of 25gm, 50gm and 100gm. The  $H_{\nu}$  was found to increase with the increase in the load from 25m to 100gm and crack occurred at higher loads as shown in Table VII.

The graph (Fig. 8) has been plotted between  $H_{\nu}$  and applied load P. The Vickers micro hardness number  $H_{\nu}$  of the crystal was calculated using the relation

$$H_v = 1.8544 \text{ P/d}^2 \text{ (kg/mm}^2\text{)}$$
 (5)

Where,  $H_{\nu}$  is the Vickers hardness number in kg/mm<sup>2</sup>, P is the applied load in kg and d is the average diagonal length of the indentation in mm. From the graph it can be observed that the hardness value increased up to 100g and the maximum hardness value was  $78.6 \text{ kg/mm}^2$  at 100 gm.

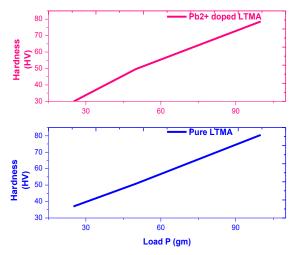


Fig. 8 Graph against Load P Vs Hardness  $(H_v)$  of Pure LTMA &  $Pb^{2+}$  doped LTMA crystal

Table-VII: Hardness values of pure LTMA and Pb<sup>2+</sup> doped LTMA crystals

40)14 221:21 21,51425			
	Hardness		
	$(\mathbf{H}_{\mathbf{V}})$		
Load P (gm)	Pure LTMA	Pb <sup>2+</sup> doped LTMA	
25	37	29.85	
50	50.8	49.65	
100	80.5	78.6	

#### IV. CONCLUSION

Novel Single crystals of Pure and Pb<sup>2+</sup> doped L-Threonine Manganese Acetate were effectively developed by slow evaporation technique. Single precious crystal X-ray diffraction examines found that the gem structure of the two gems becomes Orthorhombic and has bigger volume. The powder X-ray diffraction examination has been done the crystallinity of the contrasts between the unadulterated and Pb<sup>2+</sup> doped LTMA single crystal. EDAX study affirmed the presence of the dopant Mn in Pure LTMA and Pb in doped Crystal. UV-Vis-NIR considers found that the unadulterated and doped crystals have wide straightforwardness in the entire noticeable area and utilizing with their vitality band holes 4.3eV and 6.3eV were resolved. FTIR considers uncover that the nearness of utilitarian gatherings in the crystal. The mechanical quality of the both developed crystals was concentrated by Vicker's Micro hardness analyzer.

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